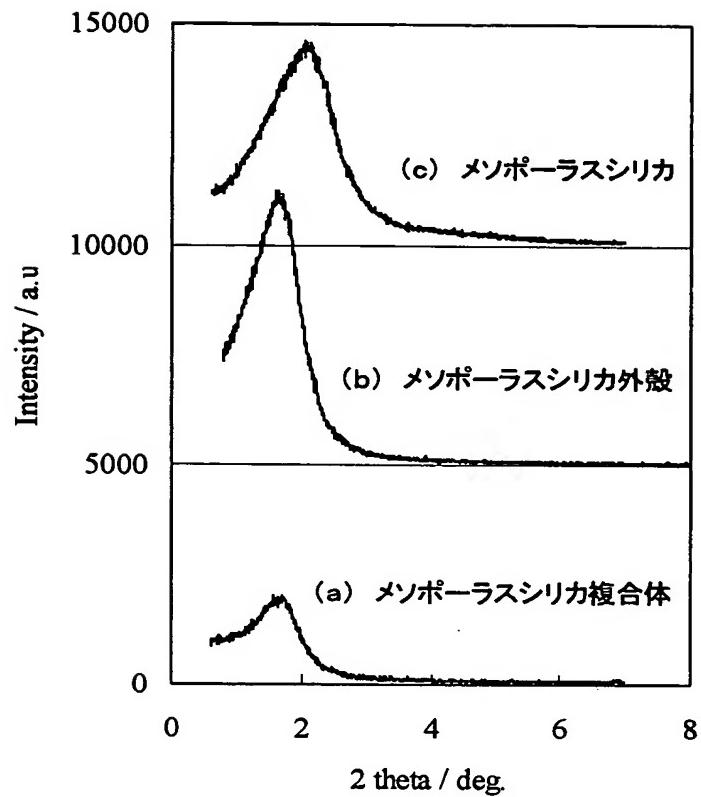


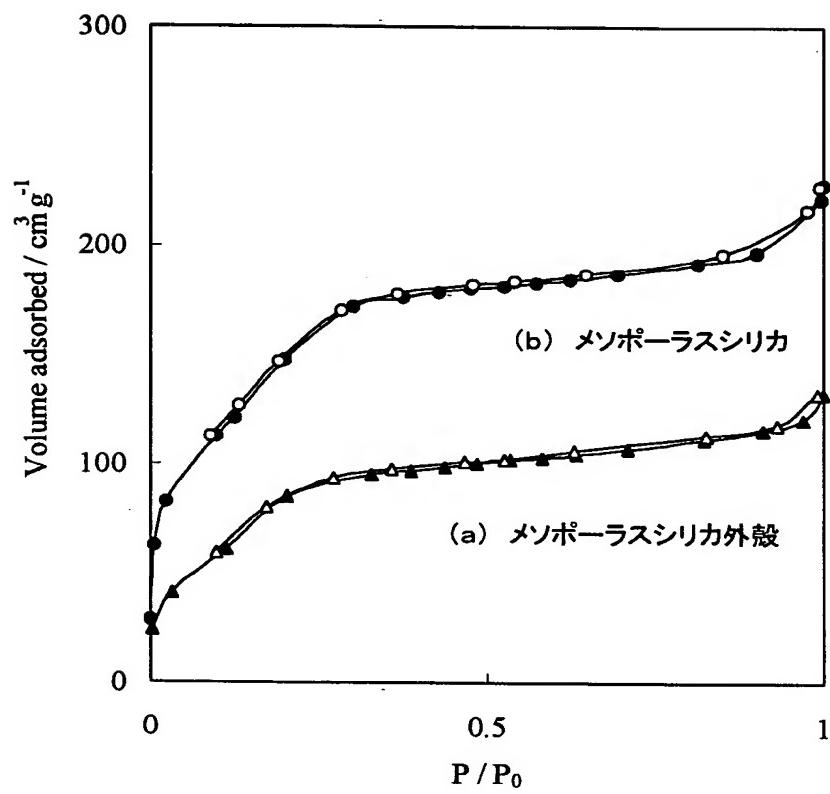
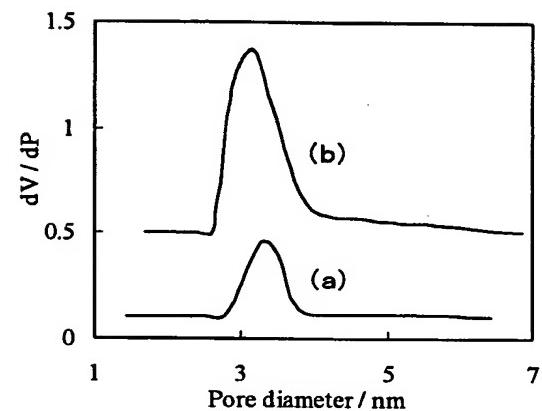
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【図2】



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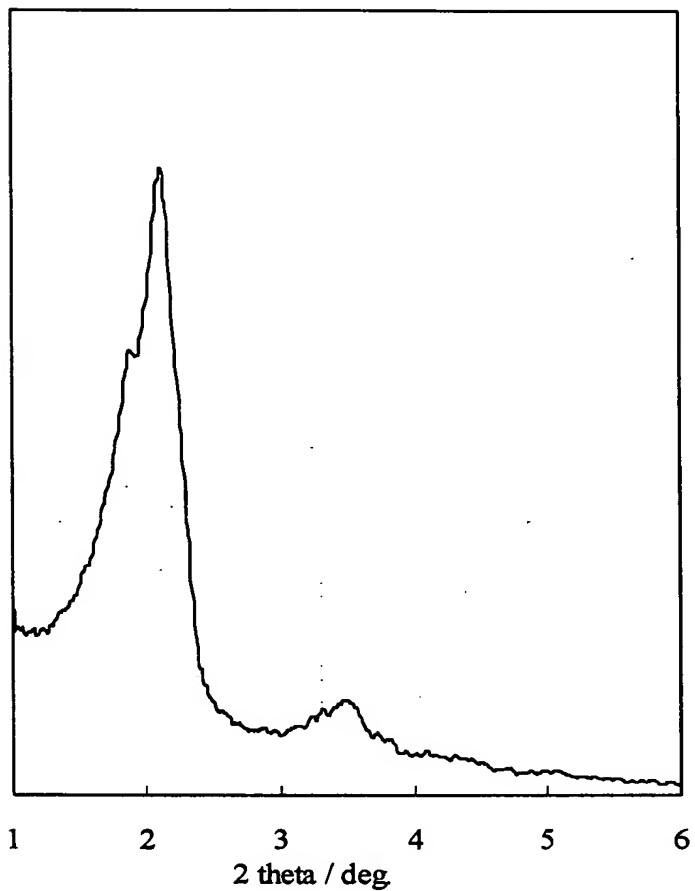
【図3】



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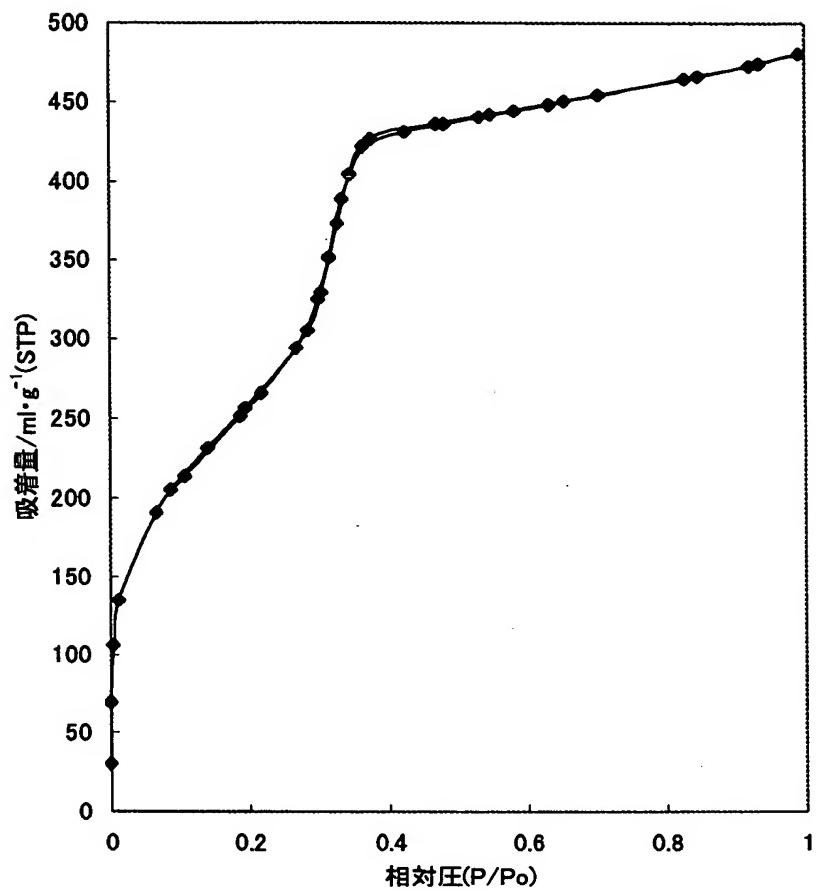
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【図4】



DF4392/US

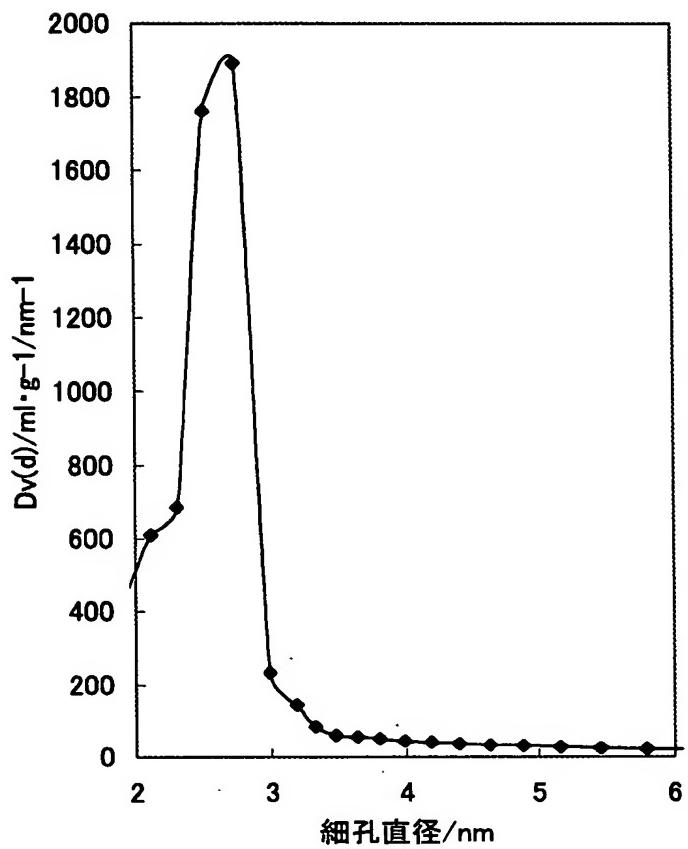
【図5】



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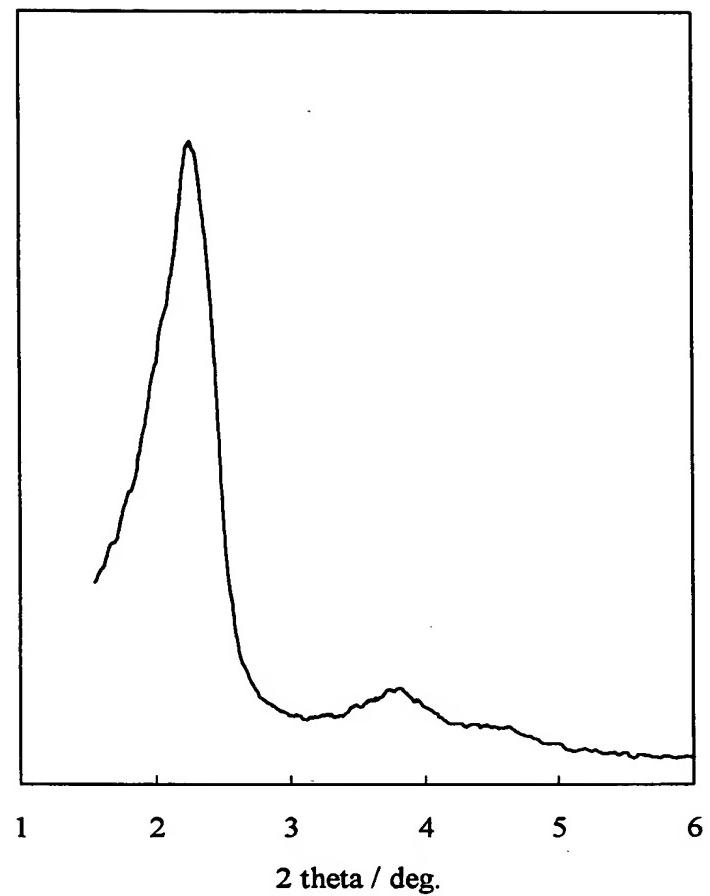
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【図6】

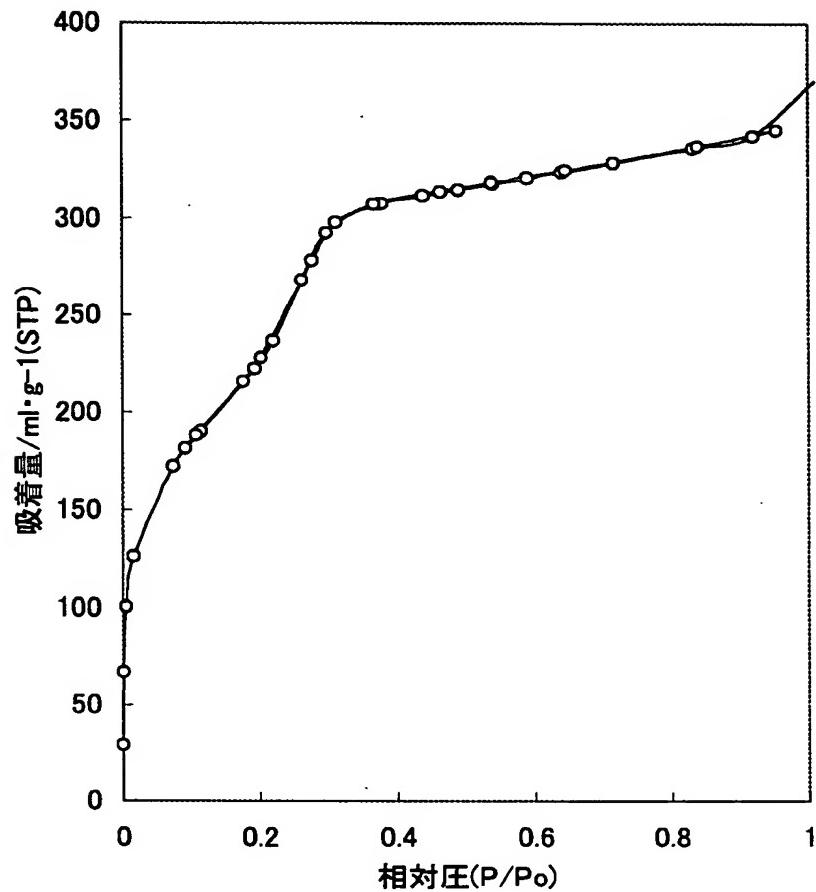


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【図7】

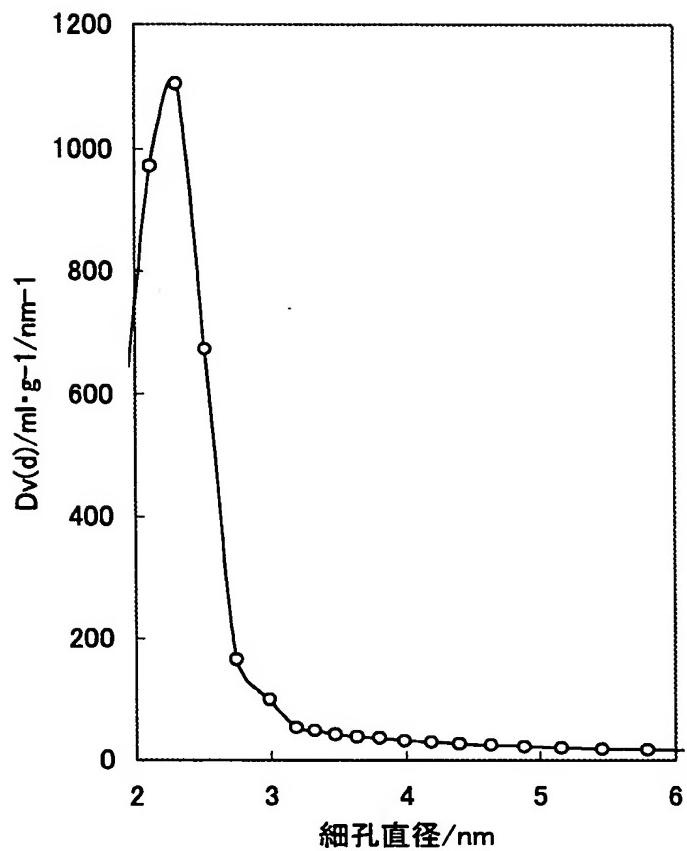


【図8】



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【図9】



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Fig. 10. Schematic illustration of the two types of amino group-anionic surfactant head group interactions: through neutralization of acid with primary aminosilane APS and double decomposition of negatively charged anionic salt surfactant with positively charged quaternized aminosilane TMAPS.

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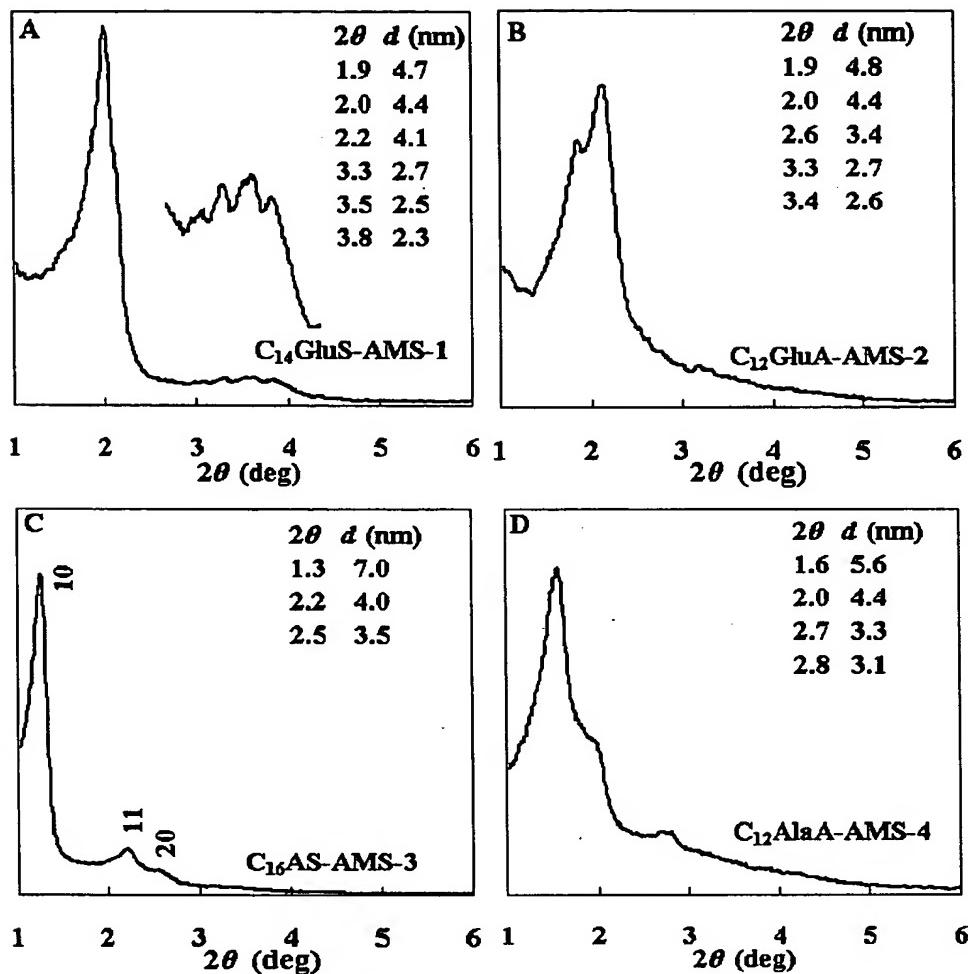


Fig. 11. XRD patterns of cacioned AMS-n mesoporous silica. The chemical mol composition of the reaction mixture was (A) C₁₄GluS-AMS-1, C₁₄GluS:TMAPS:TEOS:H₂O 1:2:10:2405 (at 100 °C for 3 d); (B) C₁₂GluA-AMS-2: C₁₂GluA:APS:TEOS:H₂O 1:2.5:18.5:1905 (at 100 °C for 2 d); (C) C₁₆AS-AMS-3: C₁₆AS:TMAPS:TEOS:H₂O 1:1:9:1544 (at 60 °C for 1 d); (D) C₁₂AlaA-AMS-4,

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C₁₂AlaA:APS:TEOS:H₂O 1:0.75:7.5:1505 (at 60 °C for 1 d). XRD patterns were recorded on an MX Labo powder diffractometer equipped with Cu K α radiation (40 kV, 20 mA) at the rate of 1.0 deg/min over the range of 1.5 – 10.0 ° (2 θ).

Supporting online materials:

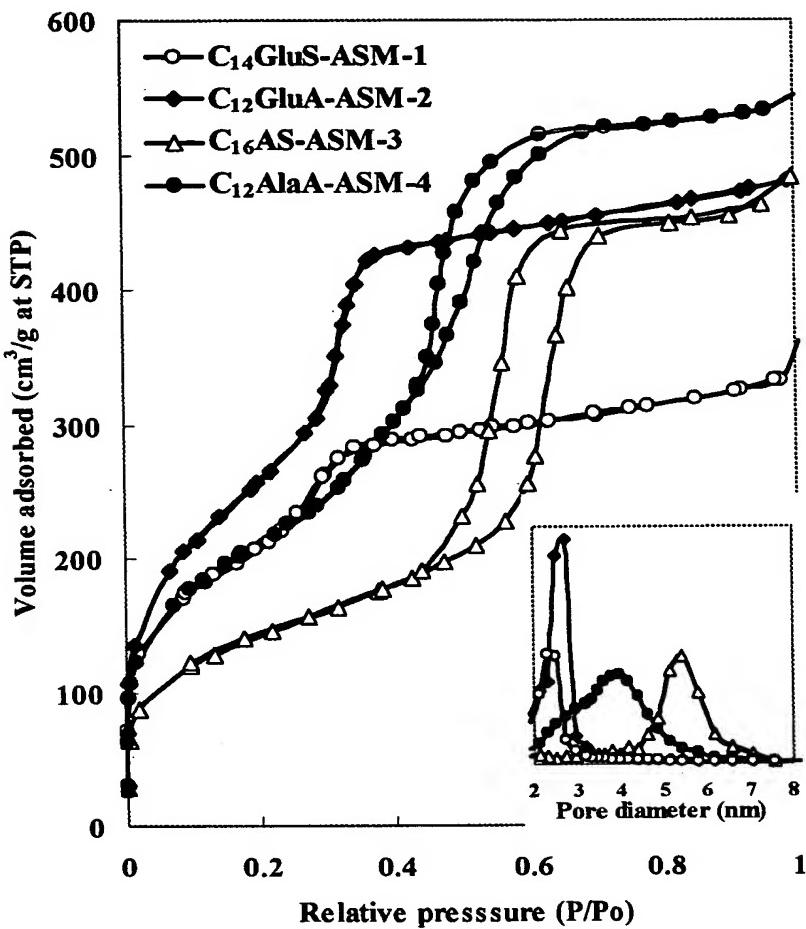


Fig. 12. N_2 adsorption-desorption isotherms and BJH pore size distributions of AMS-n mesoporous silica shown in Fig. 11. The isotherms were measured at -196°C on a Belsorp 28SA sorptionmeter.

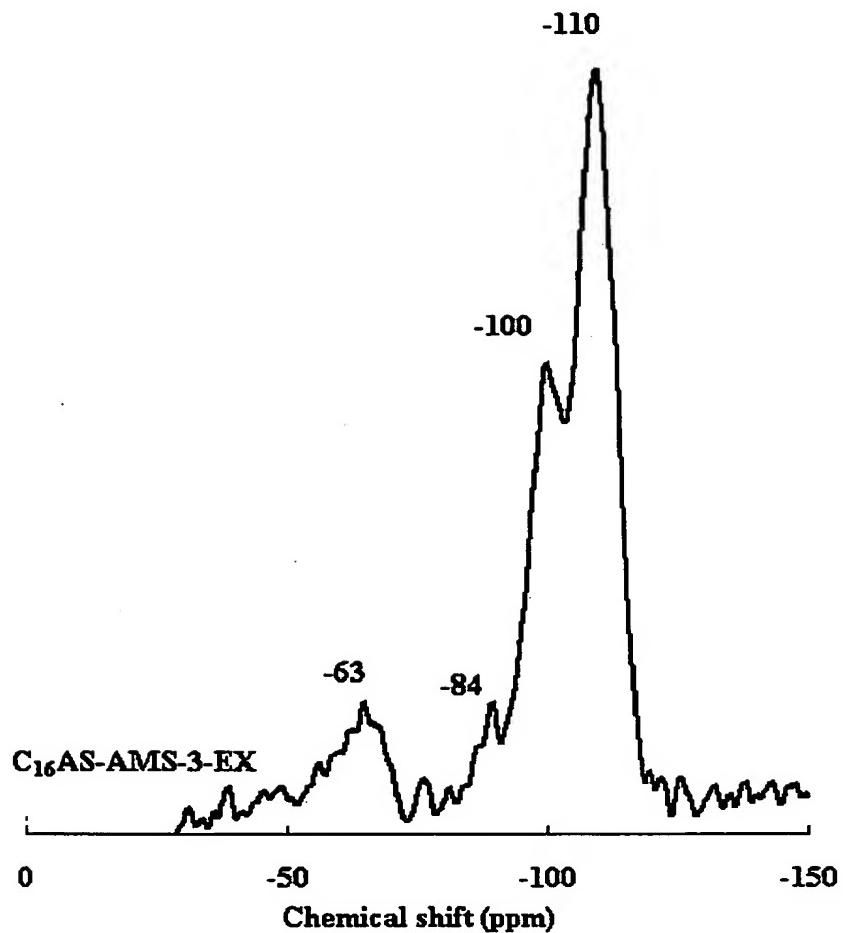


Fig. 13 shows CP ²⁹Si NMR spectra of extracted AMS-3 silica C₁₆AS-AMS-3-Ex. The spectra were collected at a JEOL-LA400WB 400 MHz spectrometer at 79.4 MHz and a sample spinning frequency of 5 kHz, respectively.